

MONTH	TOPICS	NO. OF SESSIONS
June-July	Part A: General Aptitude	2
June July	Organic Chemistry: Organic reaction mechanisms involving addition, elimination and substitution reactions with electrophilic, nucleophilic or radical species. Determination of reaction pathways.	3
	<b>Physical Chemistry</b> : Chemical kinetics: Empirical rate laws and temperature dependence; complex reactions; steady state approximation; determination of reaction mechanisms; collision and transition state theories of rate constants; unimolecular reactions; enzyme kinetics; salt effects; homogeneous catalysis; photochemical reactions.	3
	<b>Inorganic Chemistry:</b> Bioinorganic chemistry: photosystems, porphyrins, metalloenzymes, oxygen transport, electron- transfer reactions; nitrogen fixation, metal complexes in medicine.	3
	Test	1
July-Aug	Part A: General Aptitude	2
	<b>Organic Chemistry</b> : Principles of stereochemistry: Configurational and conformational isomerism in acyclic and cyclic compounds; stereogenicity, stereoselectivity, enantioselectivity, diastereoselectivity and asymmetric induction.	3
	<b>Physical Chemistry</b> : Polymer chemistry: Molar masses; kinetics of polymerization.	3
	<b>Inorganic Chemistry</b> : Structure and bonding in homo- and heteronuclear molecules, including shapes of molecules (VSEPR Theory).	3
	Test	1
Aug-Sep	<b>Organic Chemistry</b> : Structure determination of organic compounds by IR, UV-Vis, <sup>1</sup> H and <sup>13</sup> C NMR and Mass spectroscopic techniques.	3
	<b>Physical Chemistry</b> : Basic principles of quantum mechanics: Postulates; operator algebra; exactly- solvable systems: particle-in-a-box, harmonic oscillator and the hydrogen atom, including shapes of atomic orbitals; orbital and spin angular momenta; tunneling.	4
	Inorganic Chemistry:  1. Organometallic compounds: synthesis, bonding and structure, and reactivity. Organometallics in homogeneous catalysis.	4
	2. Cages and metal clusters.	



Sep-Oct	<b>Organic Chemistry</b> : Organic transformations and reagents: Functional group interconversion including oxidations and reductions; common catalysts and reagents (organic, inorganic, organometallic and enzymatic). Chemo, regio and stereoselective transformations.	1+3
	<b>Physical Chemistry</b> : 1. Electrochemistry: Nernst equation, redox systems, electrochemical cells; Debye- Hückel theory; electrolytic conductance – Kohlrausch's law and its applications; ionic equilibria; conductometric and potentiometric titrations.	4
	Inorganic Chemistry:	
	Transition elements and coordination compounds: structure, bonding	2.1
	theories, spectral and magnetic properties, reaction mechanisms.  Test	2+1
	Test	1
Oct-Nov	Organic Chemistry:  1. Benzenoid and non-benzenoid compounds — generation and reactions.	1+3
	Physical Chemistry: Chemical thermodynamics: Laws, state and path functions and their applications; thermodynamic description of various types of processes; Maxwell's relations; spontaneity and equilibria; temperature and pressure dependence of thermodynamic quantities; Le Chatelier principle; elementary description of phase transitions; phase equilibria and phase rule; thermodynamics of ideal and non-ideal gases, and solutions.	4
	<ul><li>Inorganic Chemistry: 1. Inner transition elements: spectral and magnetic properties, redox chemistry, analytical applications.</li><li>2. Concepts of acids and bases, Hard-Soft acid base concept, Nonaqueous solvents.</li></ul>	2+1
	Test	1
Nov-Dec	Organic Chemistry: 1. Pericyclic reactions; 2. Asymmetric synthesis: Chiral auxiliaries, methods of asymmetric induction — substrate, reagent and catalyst-controlled reactions; determination of enantiomeric and diastereomeric excess; enantio-discrimination. Resolution — optical and kinetic.	2+2
	Physical Chemistry:  1. Molecular spectroscopy: Rotational and vibrational spectra of diatomic molecules; electronic spectra; IR and Raman activities – selection rules; basic principles of magnetic resonance.	3
	Inorganic Chemistry:  1. Characterization of inorganic compounds by IR, Raman, NMR, EPR, Mössbauer, UV-vis, NQR, MS, electron spectroscopy and microscopic techniques.	2+2
	Test	1



## **Full Portion Mock Tests**

Month	Test
August	Mock Test 1
September	Mock Test 2
October	Mock Test 3
November	Mock Test 4
December	Mock Test 5